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Key indicators

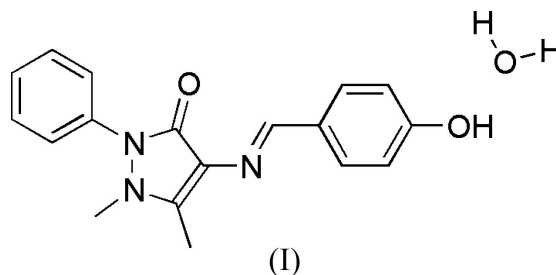
Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.038
 wR factor = 0.103
Data-to-parameter ratio = 12.4For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.4-[(4-Hydroxybenzylidene)amino]-1,5-dimethyl-2-phenyl-1*H*-pyrazol-3(2*H*)-one monohydrate

The title compound, $\text{C}_{18}\text{H}_{19}\text{N}_3\text{O}_3$, was prepared using 4-hydroxybenzaldehyde and 4-amino-1,5-dimethyl-2-phenylpyrazolidin-3-one. The crystal structure shows that the title compound includes a water molecule, which forms $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds with adjacent molecules and plays an important role in the stabilization of the crystal structure.

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Comment

In order to investigate their potential pharmacological activity (Parashar *et al.*, 1988) and photochromic properties (Hadjoudis *et al.*, 1987), a large number of metal complexes of Schiff bases have been synthesized. As part of these studies, we report the synthesis and molecular structure of the title compound, (I).



In the molecular structure, the central system (N1–N3/C3–C6/C13/O2) is planar, with an r.m.s. deviation of fitted atoms of 0.0451 Å, and the dihedral angle with the phenyl ring (C7–C12) is 89.93 (4)°. The phenol group (C15–C20/O1) is planar, with an r.m.s. deviation of fitted atoms of 0.0108 Å, and the dihedral angle with the central system is 4.82 (6)°. These values are in agreement with those in a similar system (Diao *et al.*, 2005). It is noted that the water molecules interact with pyrazolone molecules through hydrogen bonds (Table 1), which contribute to the stability of the structure in the solid state (Fig. 2).

Experimental

An anhydrous ethanol solution of 4-hydroxybenzaldehyde (1.22 g, 10 mmol) was added to an anhydrous ethanol solution of 4-amino-1,5-dimethyl-2-phenylpyrazolidin-3-one (2.03 g, 10 mmol) and the mixture was stirred at 350 K for 5 h under nitrogen. A yellow precipitate appeared which was isolated, recrystallized from ethanol and then dried *in vacuo* to give the pure compound in 83% yield. Bright-yellow single crystals suitable for X-ray analysis were obtained by slow evaporation of an ethanol solution of (I).

Crystal data

$C_{18}H_{17}N_3O_2 \cdot H_2O$
 $M_r = 325.36$
 Triclinic, $P\bar{1}$
 $a = 7.393$ (2) Å
 $b = 8.983$ (3) Å
 $c = 12.823$ (4) Å
 $\alpha = 77.510$ (4)°
 $\beta = 82.756$ (3)°
 $\gamma = 79.450$ (3)°
 $V = 814.1$ (4) Å³

$Z = 2$
 $D_x = 1.327$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 2426 reflections
 $\theta = 2.6$ – 27.7 °
 $\mu = 0.09$ mm⁻¹
 $T = 294$ (2) K
 Block, yellow
 $0.52 \times 0.46 \times 0.34$ mm

Data collection

Bruker SMART 1000 CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Bruker, 1999)
 $T_{\min} = 0.923$, $T_{\max} = 0.969$
 4423 measured reflections

2843 independent reflections
 2407 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.014$
 $\theta_{\text{max}} = 25.0$ °
 $h = -8 \rightarrow 8$
 $k = -9 \rightarrow 10$
 $l = -15 \rightarrow 13$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.103$
 $S = 1.04$
 2843 reflections
 229 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.049P)^2 + 0.1768P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.18$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.21$ e Å⁻³
 Extinction correction: SHELXL97
 Extinction coefficient: 0.082 (6)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O1–H1 \cdots O3	0.94 (2)	1.69 (2)	2.6348 (18)	178 (2)
O3–H3A \cdots O1 ⁱ	0.83 (3)	2.27 (3)	3.0182 (19)	149 (3)
O3–H3B \cdots O2 ⁱⁱ	0.91 (3)	1.81 (3)	2.7076 (18)	168 (3)

Symmetry codes: (i) $-x + 1, -y + 1, -z + 2$; (ii) $-x + 1, -y + 1, -z + 1$.

The H atoms of the hydroxyl group and water molecule were found in a difference Fourier map and refined freely with an isotropic U parameter. Other H atoms were included in calculated positions and refined as riding. C–H bond lengths and isotropic U parameters were constrained: C–H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic H atoms, and C–H = 0.96 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H atoms.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine

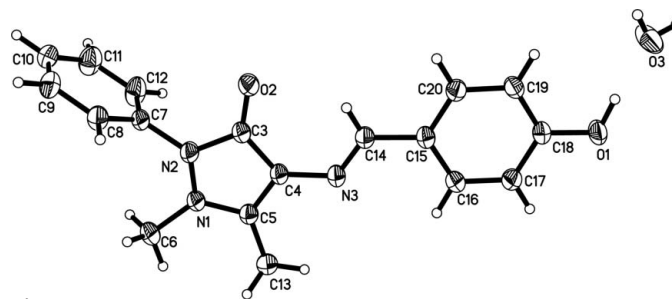


Figure 1

The structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

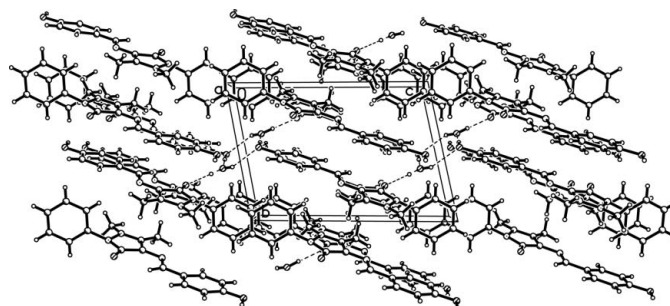


Figure 2

Intermolecular hydrogen-bonding interactions (dashed lines).

structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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